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From a methanolic extract of the needles of *Larix sibirica* (Siberian larch) by chromatography on polyamide and elution with 30% aqueous methanol we have isolated a fraction containing three substances. The chromatography of this fraction on polyamide in a nonaqueous system [chloroform-methanol (95:5)] yielded an individual compound with mp 173-175°C (methanol),  $[\alpha]_D^{20} -23.01$  [c 0.25; methanol-water (1:1)].

The UV absorption at 267 and 350 nm ( $\log \epsilon$  4.28, 4.20, 4.24, 4.33)\* and the frequency of the stretching vibrations of the C=O group in the IR spectrum (1660  $\text{cm}^{-1}$ ) show the flavonoid structure of the compound. In the NMR spectrum, two doublets ( $\delta = 6.28$  and 6.50 ppm,  $J = 2.6$  Hz) correspond to two protons at  $C_6$  and  $C_8$  of ring A. The lateral phenyl ring is para-substituted. This is shown by the bands of the deformation vibrations in the IR spectrum (810, 840  $\text{cm}^{-1}$ ) and by two doublets ( $\delta = 6.98$  and 8.11 ppm) in the NMR spectrum. The constant  $J = 9.0$  Hz corresponds to the interaction of the protons at  $C_2$ , and  $C_3$ , and those at  $C_5$ , and  $C_6$ , which are in the ortho positions with respect to one another. A signal at  $\delta$  12.68 ppm is due to the proton of a 5-OH group, the presence of which is also confirmed by a bathochromic shift of the long-wave maximum by 44 nm in the presence of  $\text{AlCl}_3$ . A broad singlet (width 60 Hz) with its center at  $\delta = 9.82$  ppm is due to the protons of the 4'-OH and 7-OH groups, leading to a displacement of the short-wave maximum by 8 nm and of the long-wave maximum by 19 nm in the UV spectrum in the presence of sodium acetate.

A signal at  $\delta = 5.49$  ppm relates to the protons of a glycosidic substituent. The  $\beta$  configuration of the glycosidic aglycone-sugar bond was shown by hydrolysis with  $\beta$ -glucosidase.

Acid hydrolysis gave the aglycone, with mp 269-272°C (aqueous methanol) with a yield of 61%, which corresponds to a monoglycoside.

In the NMR spectrum of the aglycone there is no signal at  $\delta = 5.49$  ppm. The doublets corresponding to the protons at  $C_6$  and  $C_8$ , at  $C_2'$  and  $C_6'$ , and at  $C_3'$  and  $C_5'$  agree with the corresponding signals of the protons in the spectrum of the glycoside. The appearance of a free 3-OH group on hydrolysis is satisfactorily confirmed by the increase in the bathochromic shift in the presence of  $\text{AlCl}_3$  in the UV spectrum of the aglycone as compared with the spectrum of the glycoside (55 nm). Consequently, the aglycone is 3,4',-5,7-tetrahydroxyflavone (kaempferol).

The hydrolysate was shown to contain glucose by paper chromatography [butanol-pyridine- $\text{H}_2\text{O}$  (10:3:3)].

The experimental results that we have obtained permit the compound isolated to be characterized as kaempferol 3-glucoside (astragalín), which has been found in the flowers of *Aesculus hippocastanum* L. [1].

The NMR spectra were taken by V. K. Voronov on a BS487B spectrometer. Dimethyl sulfoxide was used as solvent and as internal standard. The values of the chemical shifts are given relative to the signal of hexamethyldisiloxane.

\*As in Russian original - Publisher.

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#### LITERATURE CITED

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